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IA H 1		The Periodic Table															VIIIA He 2
1.01	IIA	ř.										IIIA	IVA	VA	VIA	VIIA	4.00
Li	Be											B 5	C	N	0	F	Ne
3 6.94	4 9.01											5 10.81	6 12.01	7	8 16.00	9 19.00	10 20.18
Na	Mg	2										AL	Si	P	S	CI	Ar
11	12											13	14	15	16	17	18
22.99	24.31	IIIB	IVB	VB	VIB	VIIB	VIIIB	VIIIB	VIIIB	IB.	IIB	26.98	28.09	30.97	32.07	35.45	39.95
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
39.10	40.08	44.96	47.88	50.94	52.00	54.94	55.85	58.93	58.69	63.55	65.39	69.72	72.61	74.92	78.96	79.90	83.80
Rb	Sr	Y	Zr	Nb	Мо	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te		Xe
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
85.47	87.62	88.91	91.22	92.91	95.94	(97.9)	101.07	102.91	106.42	107.87	112.41	114.82	118.71	121.76	127.60	126.90	131.29
Cs	Ba	La	Hf	Та	W	Re	Os	Ir	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
	137.33	138.91	178.49	180.95	183.85	186.21	190.2	192.22	195.08	197.97	200.59	204.38	207.2	208.98	(209)	(210)	(222)
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Uub	Uut	Uuq	Uup			
87	88	89	104	105	106	107	108	109	110	111	112	113	114	115			
223.02	226.03	227.03	(261)	(262)	263)	(262)	(265)	(266)	(271)	(272)	(285)	(284)	(289)	(288)			
				Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
				58	59	60	61	62	63	64	65	66	67	68	69	70	71
				140.12		144.24	(145)		152.97	2	158.93					173.04	174.97
				Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
				90	91	92	93	94	95	96	97	98	99	100	101	102	103
				232.04	231.04	238.03	237.05	(240)	243.06	(247)	(248)	(251)	252.08	257.10	(257)	259.10	262.11

Some Useful Formula and Constants:

$$K_{p} = K_{c} (RT)^{\Delta n}$$

$$pH + pOH = 14 @ 25^{\circ}C$$

$$K_{a} K_{b} = 1 \times 10^{-14} @ 25^{\circ}C$$

$$\int_{n} \frac{K_{2}}{K_{1}} = -\frac{\Delta H}{R}^{\circ} \left(\frac{1}{T_{2}} - \frac{1}{T_{1}}\right)$$

$$K_{w} = 1 \times 10^{-14} @ 25^{\circ}C$$